

Micromachined gas sensor and integrated optical circuit



Alberto Lamagna & Maximiliano Fischer

MEMS Division

Physics Dept.

CAC-CNEA

MEMS Group at Physics Dept.

Comisión Nacional de Energía Atómica (CNEA)

A. Lamagna (Physicist)
A. Boselli (Physicist)
D. Rodriguez (Physicist; Ph. D student)
P. Perillo (Chemist, Ms. Sc.Mat.)
G. Molinaro (Electronic Eng. Student & Technician)
M. Fischer (Aero, Ms.MEMS)
C. Ruiz (Engineer)
F. Nesprias (Physicist, Ms. Mat. Sc. student)
G. Redelico (Electronic Eng. Student & Technician)
L. Pepe (Electronic Eng. Student & Technician)



Semiconductor Lab., CITEFA,

J. Gimenez (Electronic Eng.) R. Marabini, (Electronic Eng.), C. Arrieta(Electronic Eng.)
H. Lacomí (Electronic Eng. Student)

Technicians : C. Gillari, D. Valerio, L. Alaniz, C. Gasulla,

School of Science and Technology , Universidad Nacional de San Martín (UNSAM)

S. Reich (Physicist)



FINANCING:

***Comision Nac. Actividades Espaciales (CoNAE)**

***JPA SA**

er Ablation Lab.– Engineer Faculty, Universidad de Buenos Aires

S. Duhalde (Physicist)
F. Vignolo (Physicist)

Other collaborations:

at CNEA:

N. Scoccola, G. Soler Illia
J. Magallanes, H. Pastoriza,
C. Romero, D. Shalom

Institute of Tecnology for Food (INTA) & Universidad Catolica Argentina



Italy

Istituto Microelectronica e Microsistemi CNR

L. Dori, G. Cardinali,
L. Correra, S. Nicoletti

Dipartimento Elettronica, Univ. Di Roma:C. Di Natale

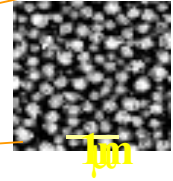


Micromachined Gas Sensors with nanostructures SnO₂ films

Outline

- * Gas sensor Architecture
 - * Tecnological Process
 - * Deposition Thin Film Methods
- * Structural Characterization of the film
- Functional Characterization of the sensor
 - * Applications of MEMS gas sensors
 - gas monitors & e-noses
- Summary and perspectives

Sensor microstructure



1 mm

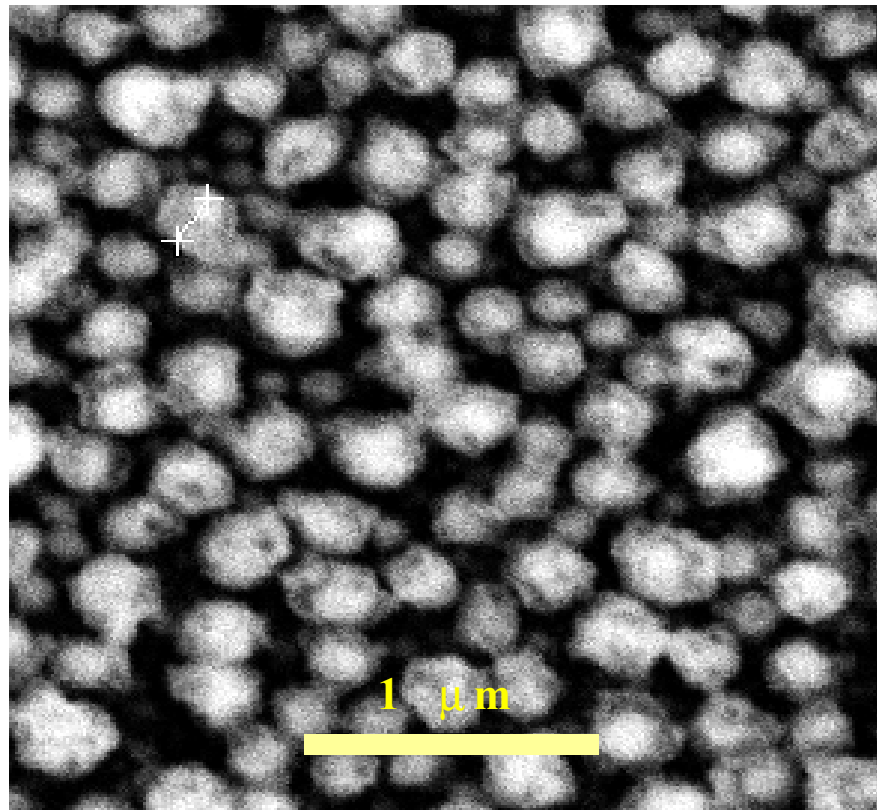
- Semiconductor chemical sensors: transduce chemical composition change in conductivity change
- Constant voltage (1 Volt) → current change

- *IC compatible fabrication process;*
- *Low thermal conductivity of the $\text{Si}_3\text{N}_4/\text{SiO}_2$ membrane;*
- *Reduced thermal mass of the device;*

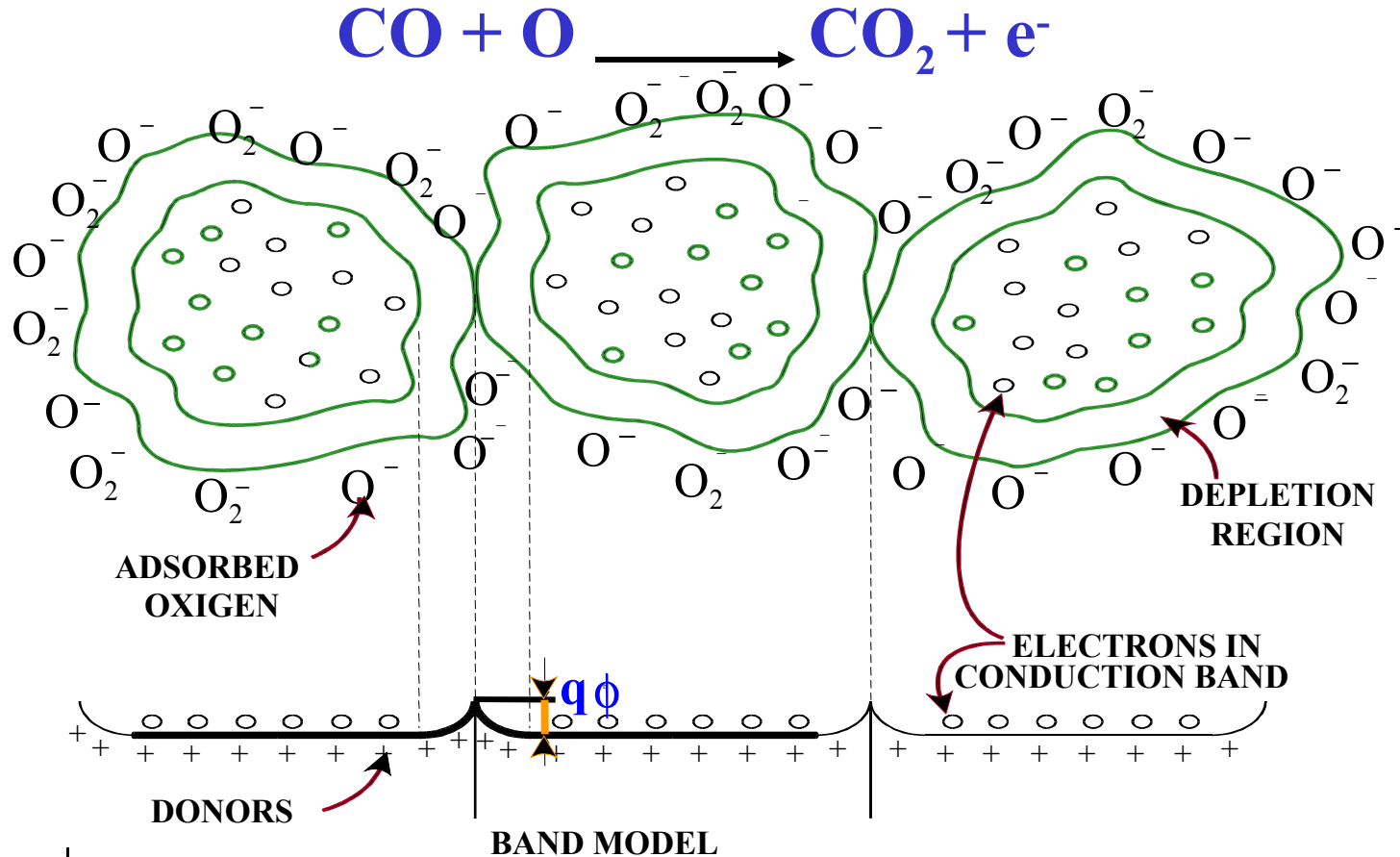
Rise Time and Recover Time \cong 30 ms

(drawing: courtesy of S. Nicoletti, IMM-CNR Bologna Italy)

SEM micrograph of 200nm thick SnO₂ film

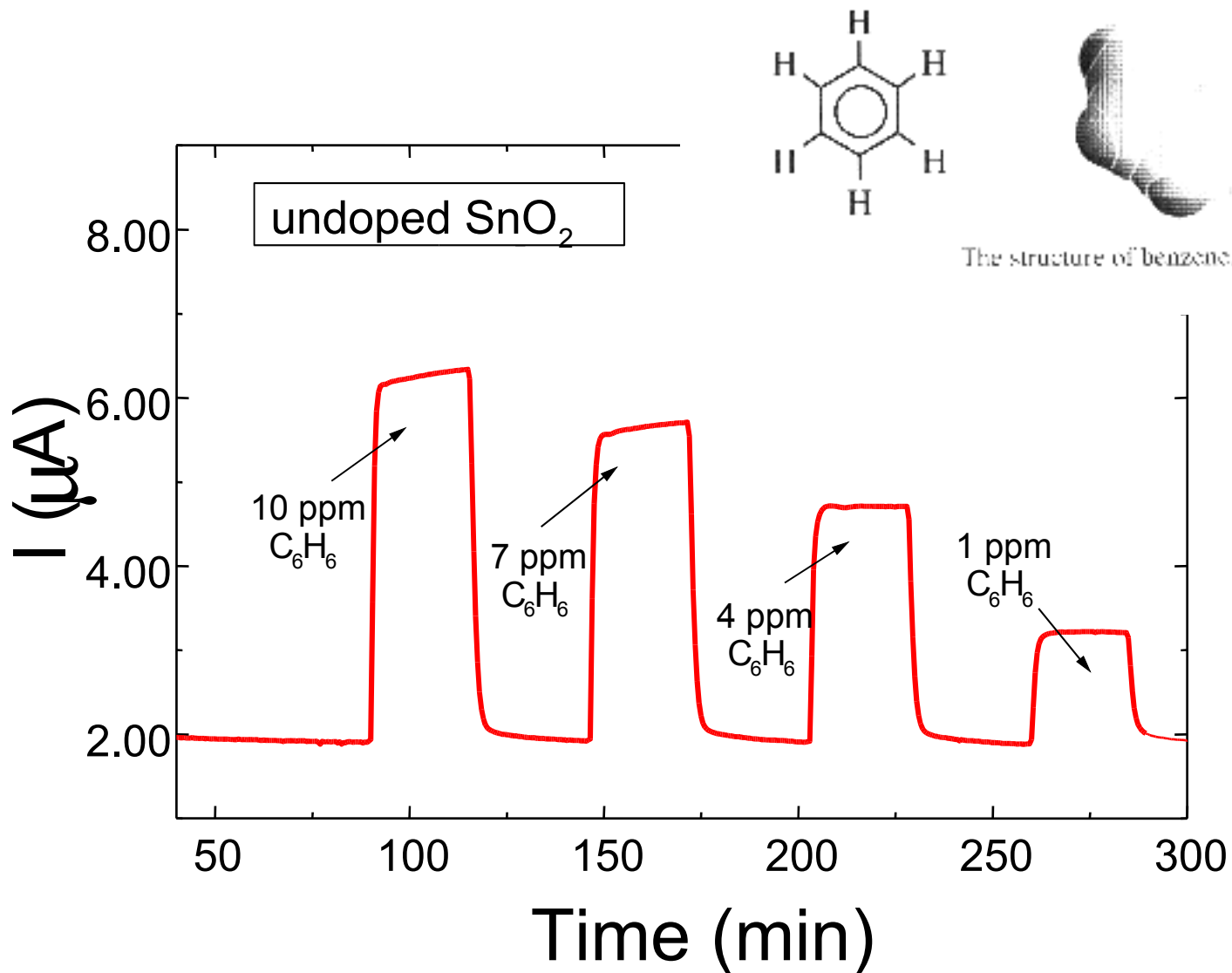


Conduction model for polycrystalline semiconductors



- The sensitive film is constituted of a nanostructured polycrystalline grain-structured Tin Oxide layer (SnO_2)
- Gases interact with the oxygen adsorbed on the film surface;
Red/Ox Reactions \Leftrightarrow Variation on the conductivity of the film

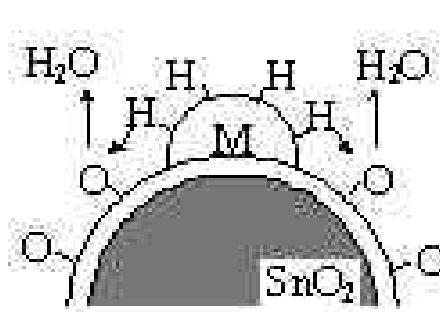
Conductance change of 200 nm thick undoped SnO_2 film when exposed to different benzene concentrations (1÷10 ppm range).



Enhanced selectivity to different gases

- device operates via resistivity changes
- gas molecules carry out a reduction/oxidation process on grains surfaces, increasing/ reducing the resistance of the film
- device operates at 200 to 500 °C
- selectivity could be modified for a particular gas species doping the metallic oxide with different elements (Pt , Pd , Au, V, etc...)
- Oxygen presence required for normal operation

Sensitivity increase by adding a catalyst

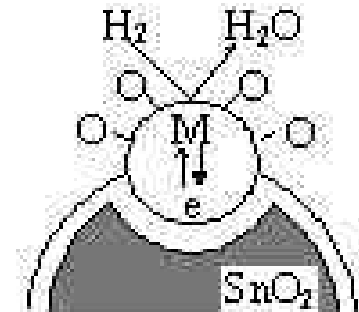


role of the metal

activation and spill-over of the gas

change of the concentration of the adsorbed oxygen

Pt



donor and acceptor of electrons

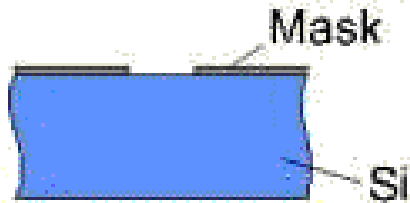
metal oxidation change of state

Ag₂O, PdO

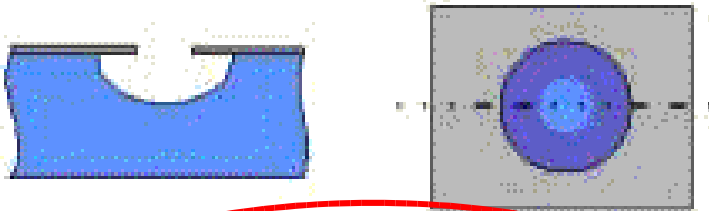
Device Structure

Bulk Micromachining

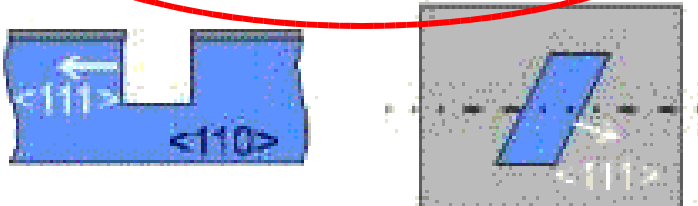
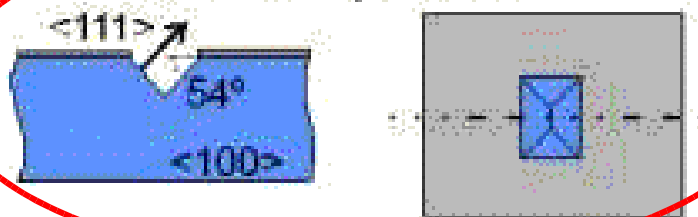
- Isotropic chemical etching
- Chemical Reactives for Si
anisotropic: KOH, EDP
- etch stop is done using a film highly doped or Si_3N_4



Isotropic etch



Anisotropic etch

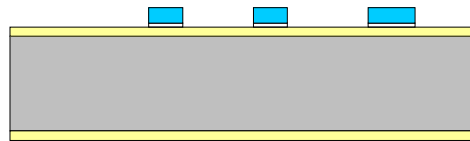




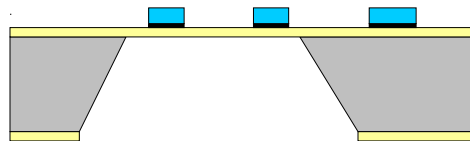
Deposition of 200 nm of Si_3N_4 on both sides of the silicon wafer



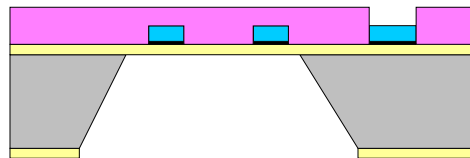
Evaporation of Ti thin film for adhesion



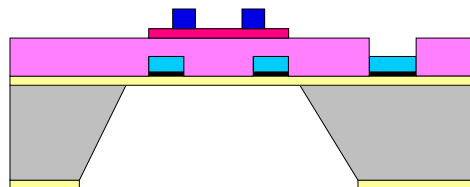
Evaporation and patterning of 300 nm Pt films for the heating resistor and the temperature sensor



Micromachining of the back of the Si wafer, in KOH - DI water solution at 80°C , leaving a freestanding Si_3N_4 membrane



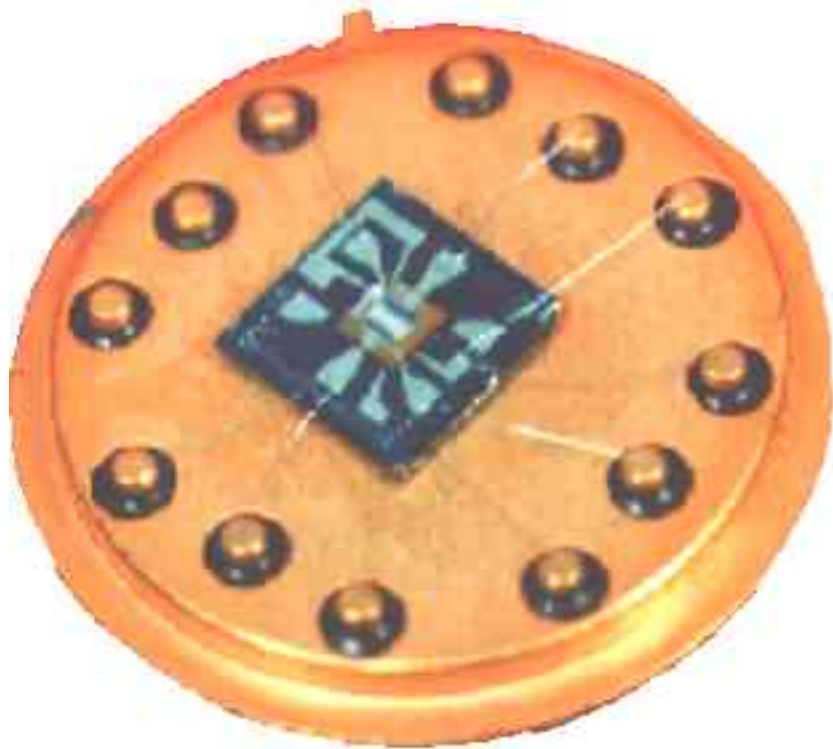
SOG film spinning for insulation of sensitive film and heater



Deposition of SnO_2 and contact pads

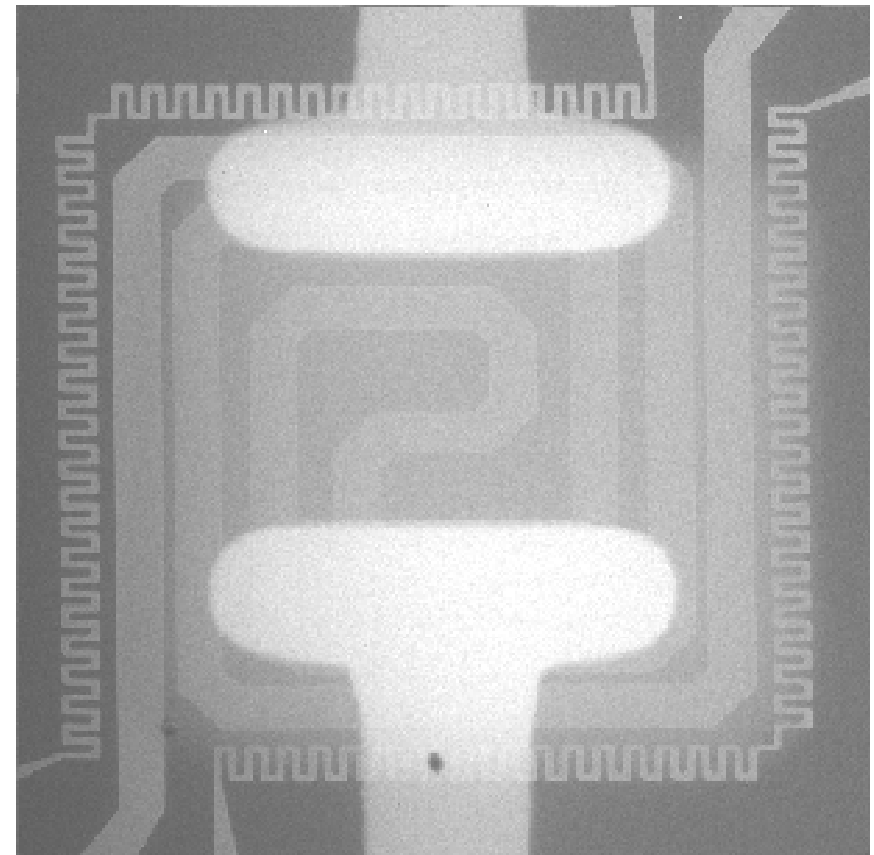
***SEM IMAGE OF
SENSOR DEVICE
SHOWING Si_3N_4
MEMBRANE***

12 lead TO-8 package



Sensor packaging

0.8 mm

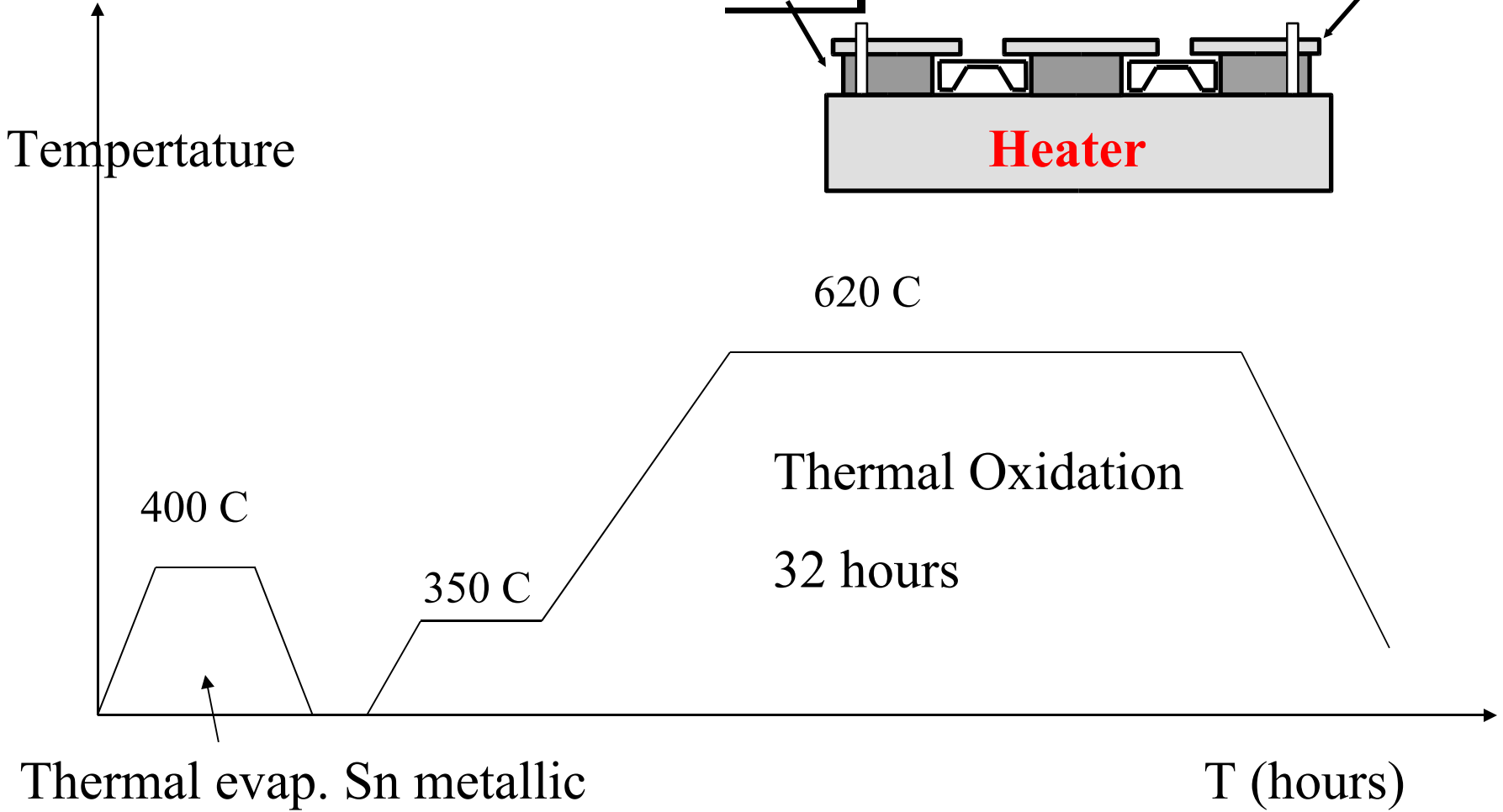
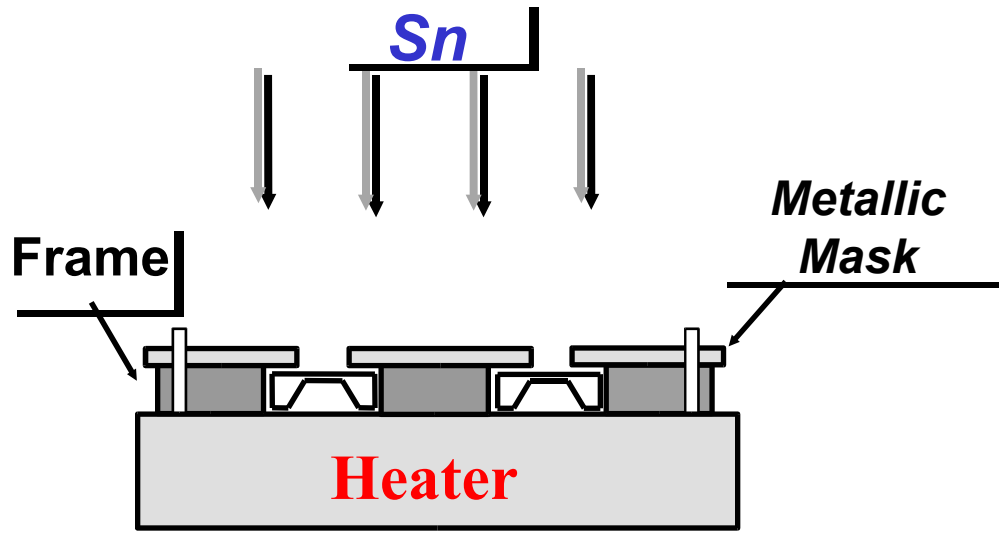


Methods for SnO₂ thin film deposition on nitride membrane

evaporation + thermal evaporation (RTGO)

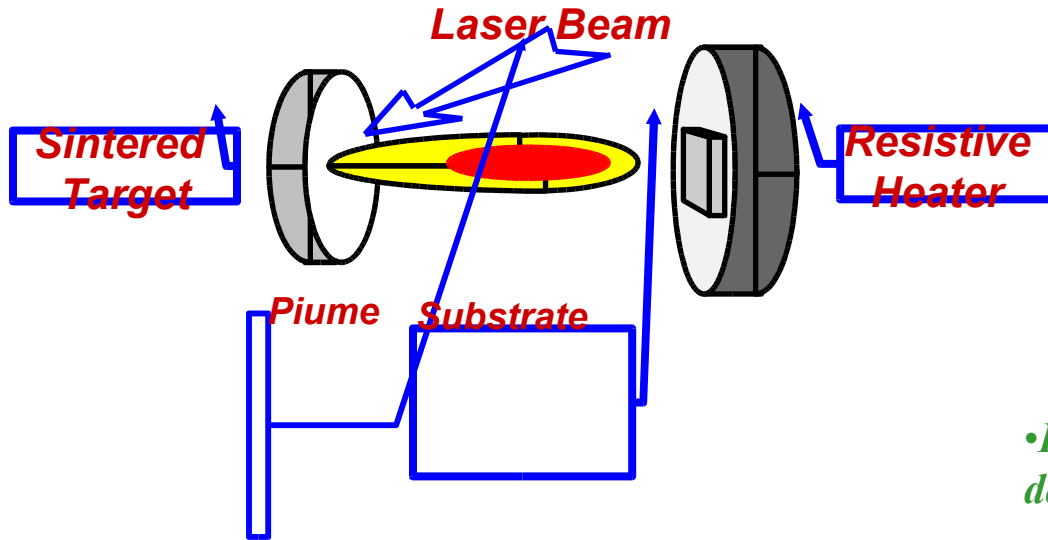
pulsed laser depositon (PLD)

Rheotaxial Growth and Thermal Oxidation (RGTO)

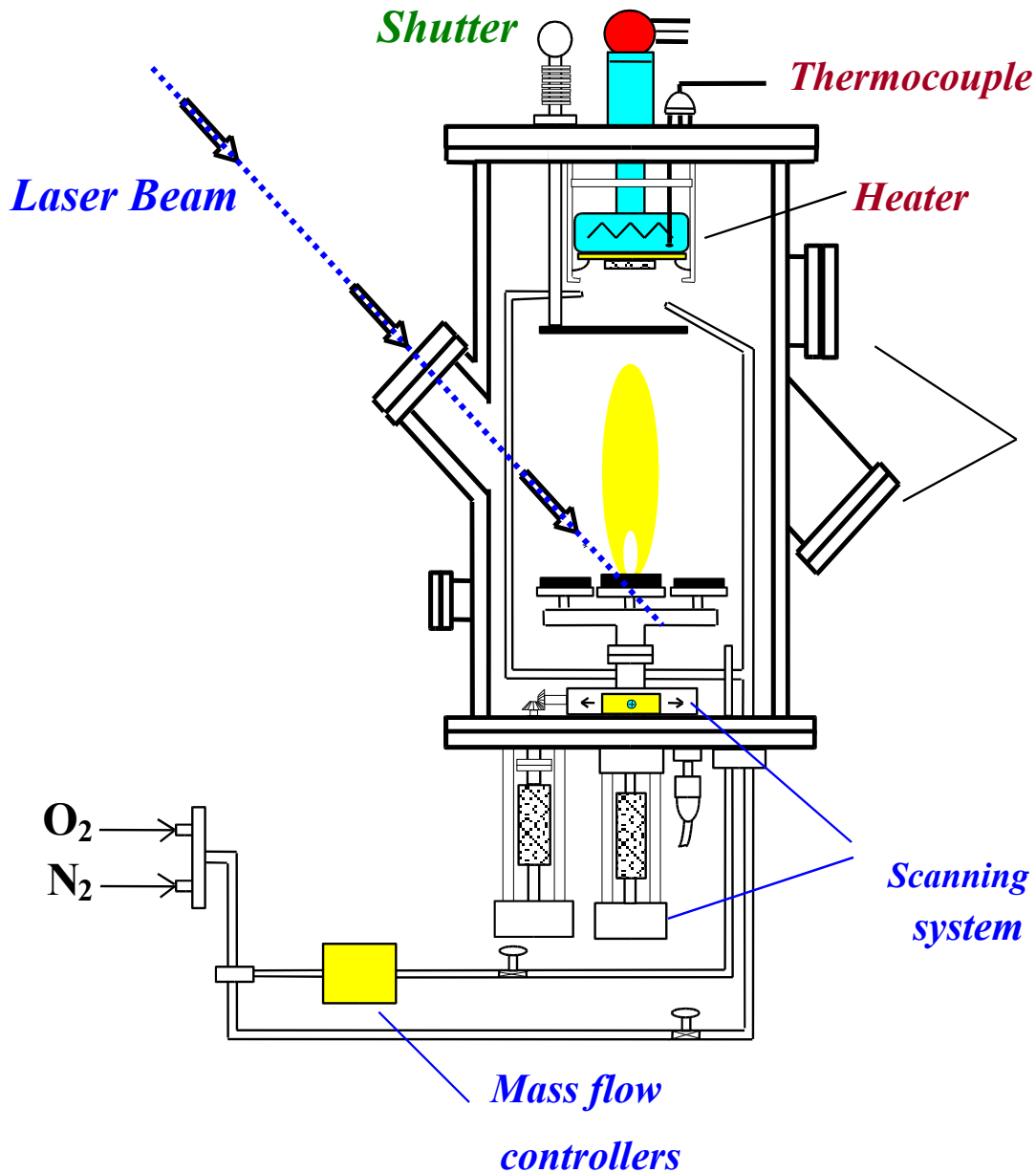


.Demare, R.Sanjines, *Gas Sensor*, Ed. by Sberveglieri, Kluwer Acad. Publ., Netherlands, 1992.

Pulsed Laser Ablation Deposition Scheme



- *PLD is a flexible technique for thin film deposition*
- *A wide range on evaporation conditions could be used (laser fluence, gas pressure, reactive environments,)*
- *Different materials and various thin films could be deposited with multitarget system;*
- *In situ oxidation*



Pulsed Laser Ablation Deposition System

**Deposition Chamber
and Energy Source are
completely decoupled**

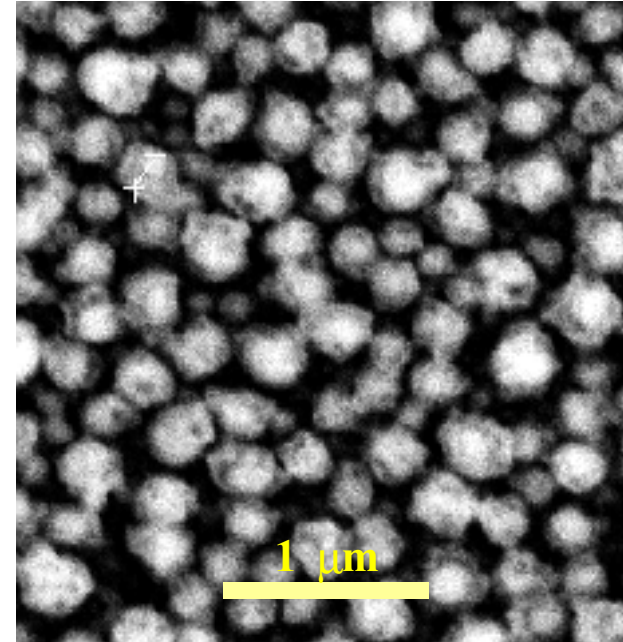
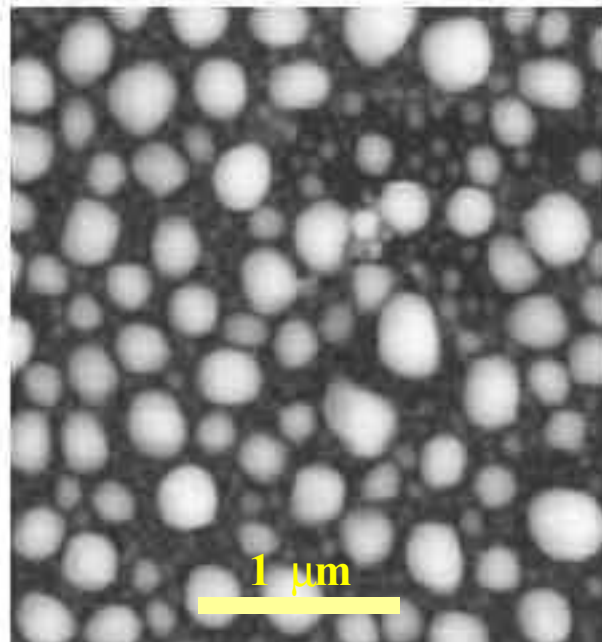
**This allows a wide range
of deposition conditions
(Laser Fluence, Gas
Pressure, Reactive or
Inert environments,)**

**MultiTarget facility
available**

Structural Characterization of the SnO₂ thin films

Thin Film before and after thermal oxidation

Rheotaxial Growth and Thermal Oxidation (RGTO)

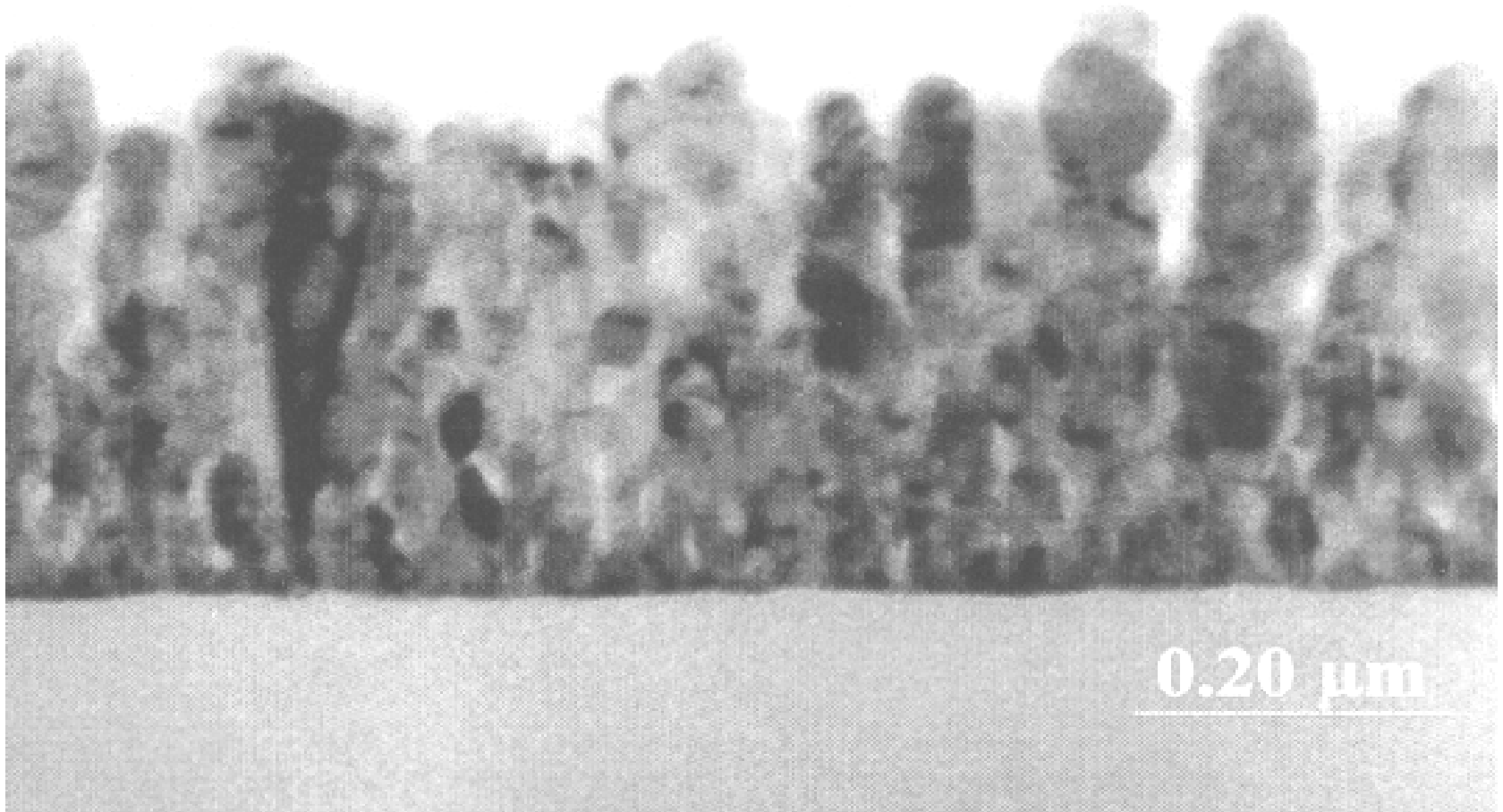


Left: SEM micrograph showing metallic drops of Sn before thermal oxidation.

Right: SEM micrograph showing SnO₂ grains after oxidation

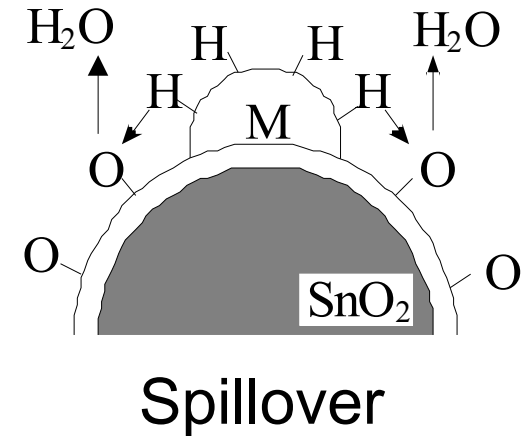
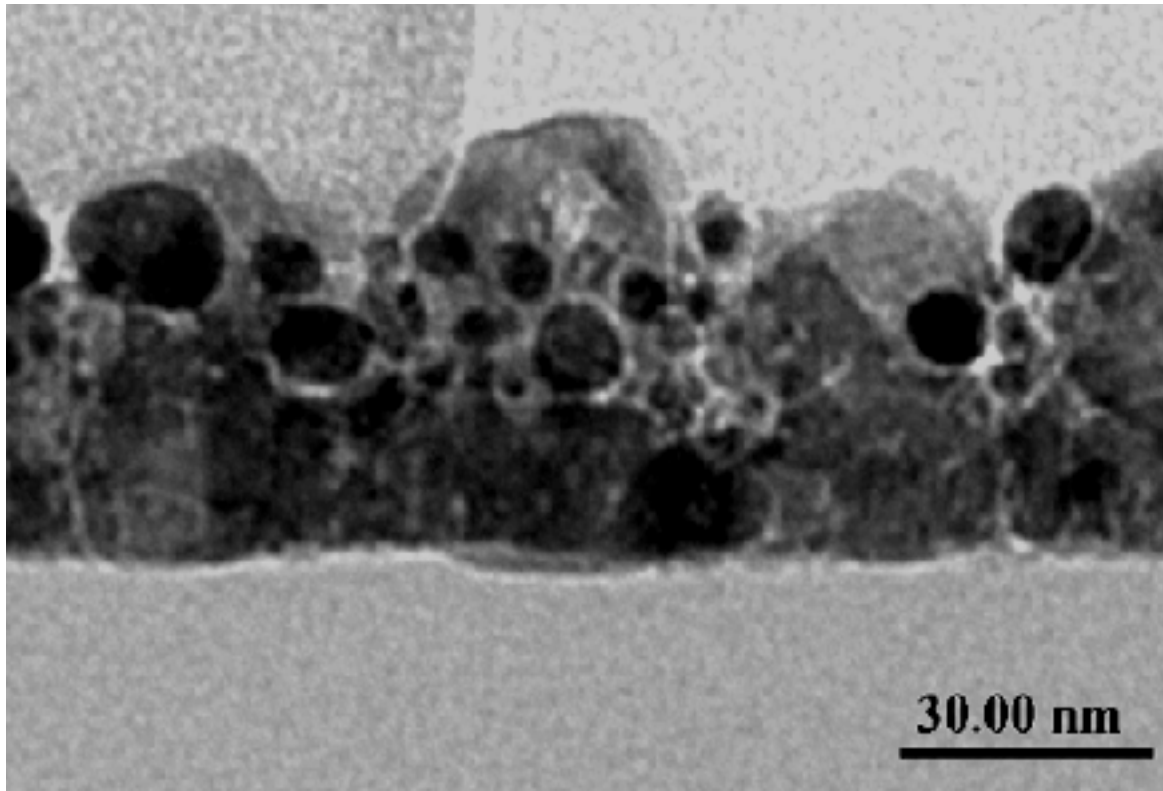
30 nm + 1200 nm steps better relation Surface/volume

TEM Micrograph of the SnO₂ film



TEM micrograph showing grains with high porosity and high specific surface

Doping of SnO₂ thin film surface using catalysts

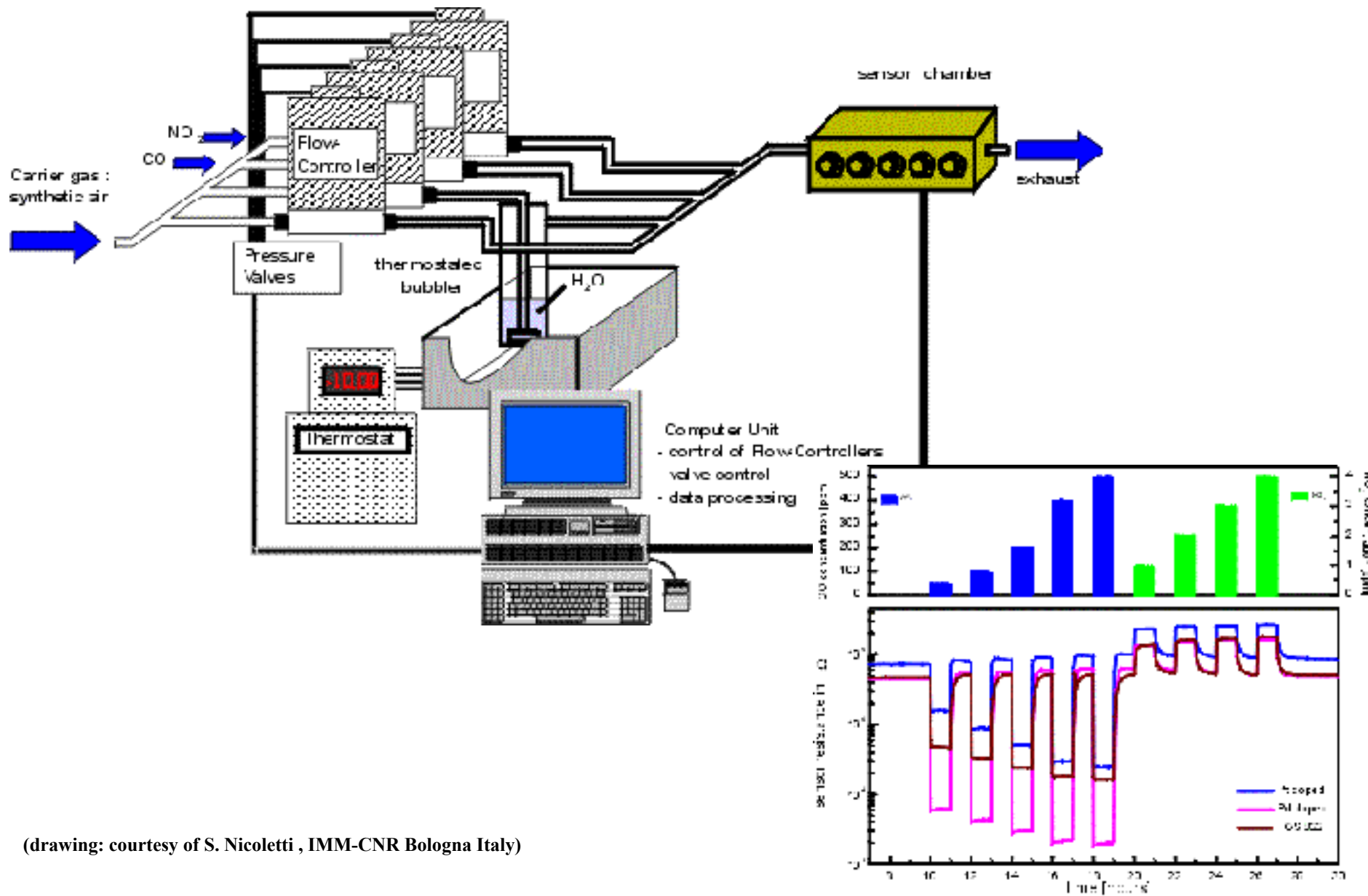


TEM micrograph showing doping material nano-drops for enhancing selectivity to certain gas species

Full oxidation of the SnO₂ thin film

Functional Characterization of sensors

gas mixer/chamber with humidity control system for quantitative sensor characterization

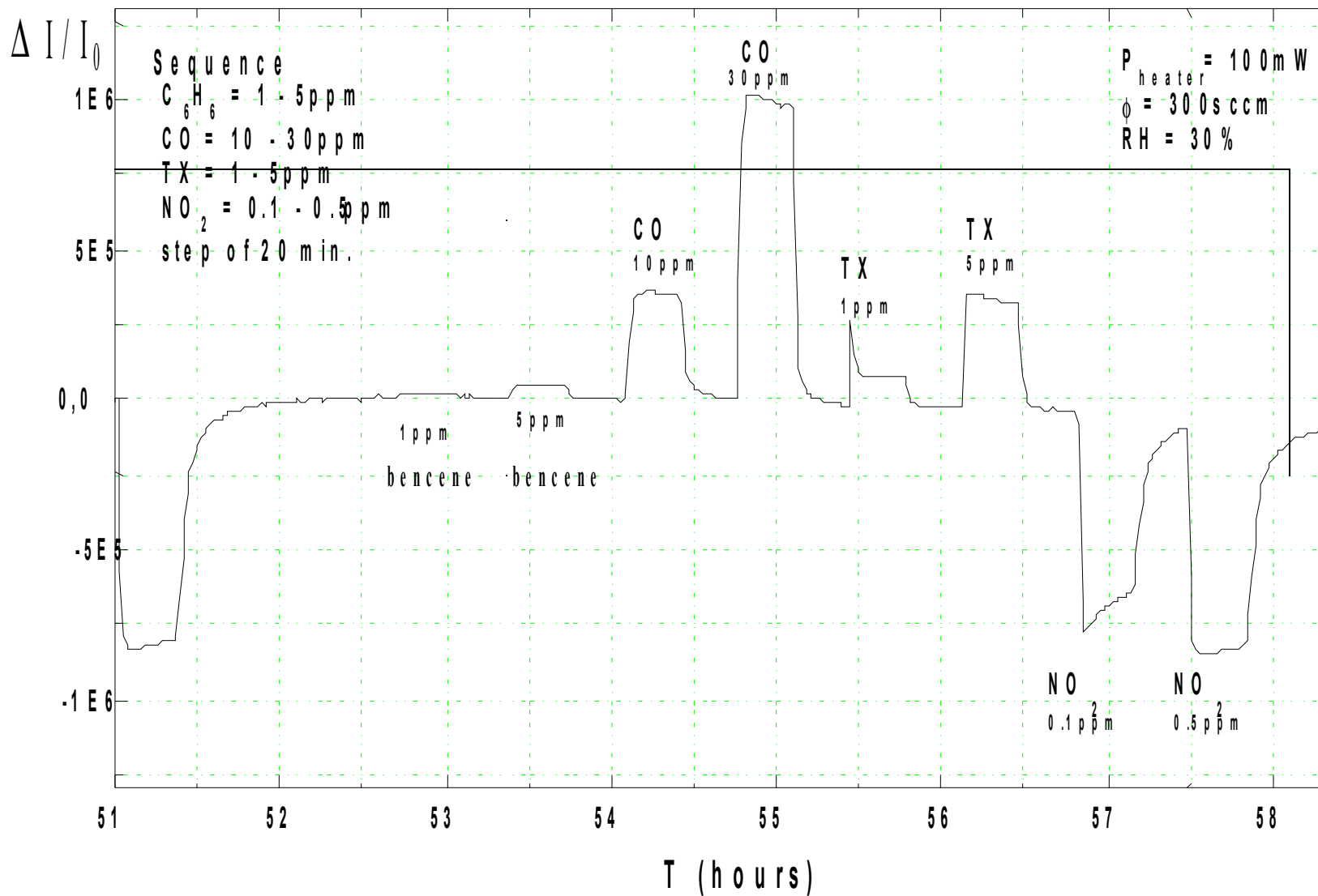


(drawing: courtesy of S. Nicoletti, IMM-CNR Bologna Italy)

**gas mixer/chamber
with humidity control system
for quantitative sensor
characterization**



R G T O _ 2 -



Functional characterization of an undoped SnO₂ sensor

Why MEMS gas sensors ?

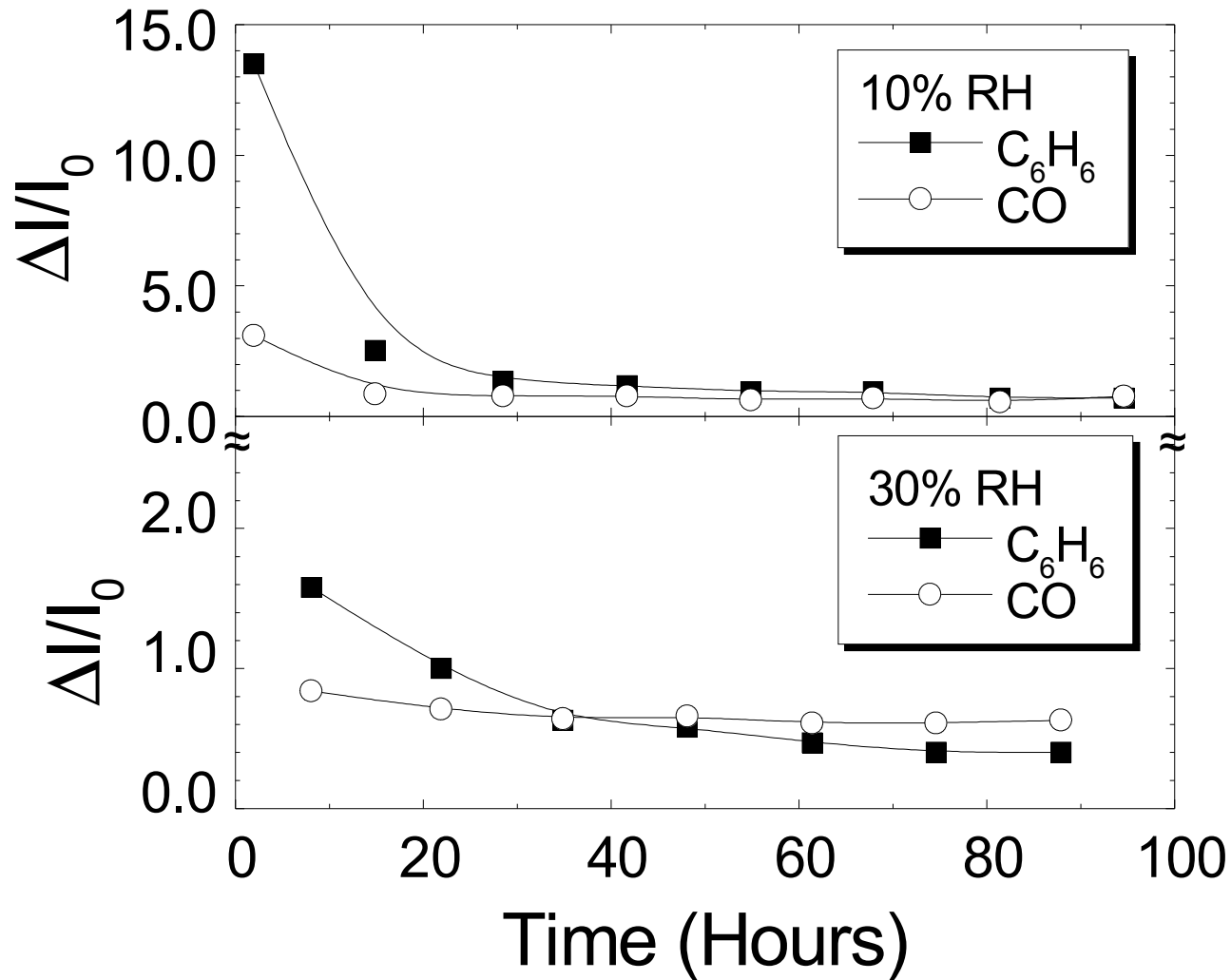
- *IC compatible fabrication process*
- *Low thermal conductivity of the $\text{Si}_3\text{N}_4/\text{SiO}_2$ membrane => low power consumption*
- *Thin film => Reduced thermal device mass => Rise Time and Recover Time $\cong 30$ ms*
- *HAND HELD DEVICES !*

Problems to be solved

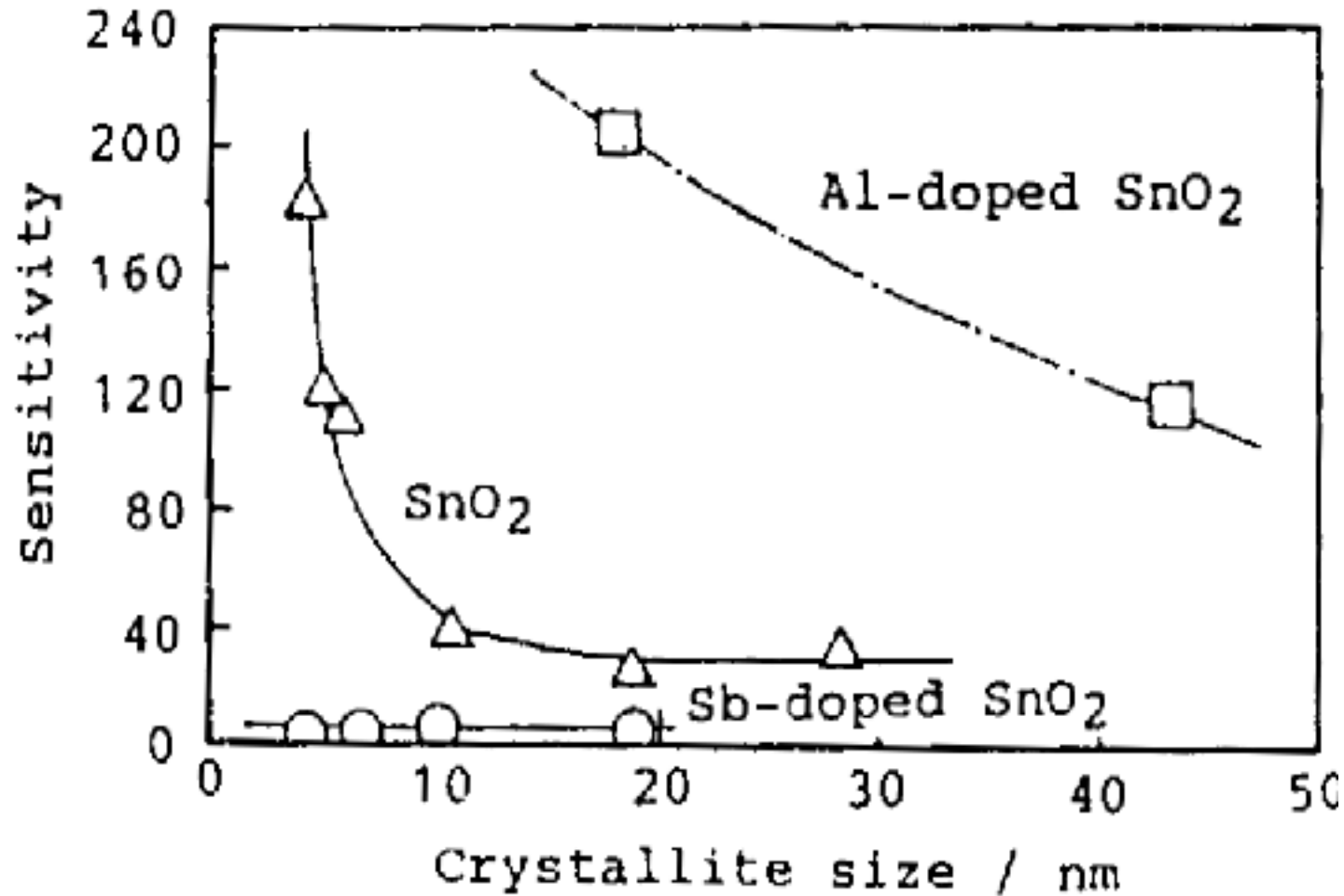
The loose of sensibility with time , the so called “aging” of the sensor

All technologies of gas sensors has this problem

$\Delta I/I_0$ as a function of the ageing time



Effects of grain size and geometry of grain connections (SnO_2)



Sensitivity is highly dependent on grain size

Applications

gas monitor

electronic noses

SMOG

Smart air pollution Monitoring networks

Research project funded by the European Commission
under the ESPRIT programme

The monitoring unit is based on an array of solid-state sensors achieved by using innovative materials and technologies for measuring **Carbon monoxide, Nitrogen dioxide, Ozone, Benzene**

Pollutant	Minimum detectable level	Pre-alarm limit	Alarm limit
Carbon Monoxide CO	1	13	26
Nitrogen Dioxide NO ₂	0.05	0.1	0.2
Ozone O ₃	0.01	0.1	0.2
Benzene C ₆ H ₆	0.5 ²	0.05	0.01

¹ Values indicated in ppm

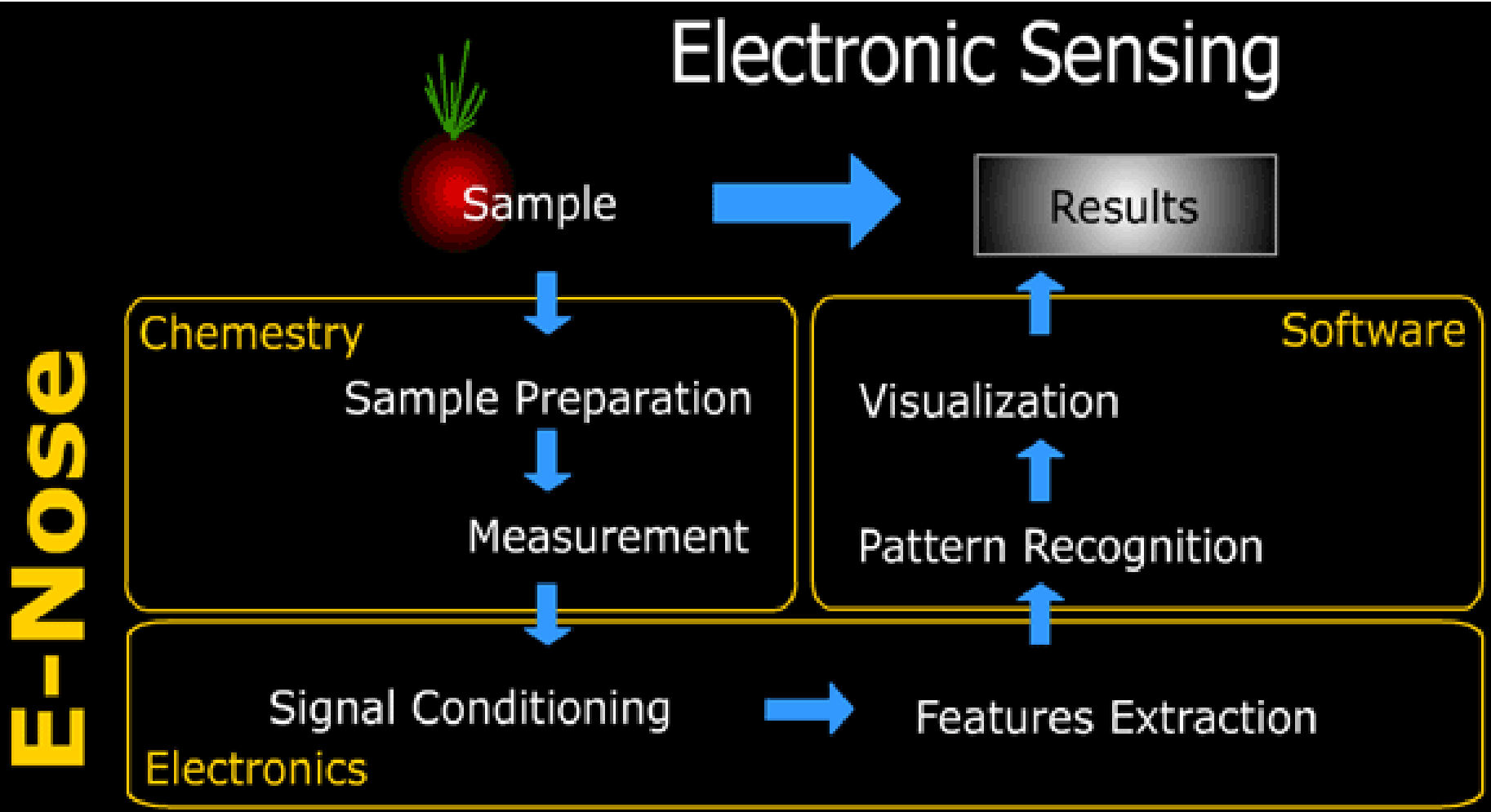
² A pre-concentrator is used to increase sensor sensitivity



Features

- * small size for enhanced compatibility with a multitude of applications;
- * power saving;
- * elements made on substrata of microprocessed Si in order to obtain miniaturized thin-film sensors with low power consumption rates at the maximum running temperatures of the sensors themselves;
- * solid-state sensors made with technologies compatible with the typical production processes of integrated circuit technologies; such sensors feature excellent stability in time and excellent reproducibility of the characteristic response to the pollutants of interest;
- * miniaturized gas chromatograph for measuring BTX
- * development of a miniaturized air sampling system using a pneumatic board, microvalves and micropumps;
- * the ability to identify and measure the concentration of the pollutant to be monitored taking into account the deviation in time typical of these sensors by means of pattern recognition techniques (neural networks, fuzzy logic);
- * easy to exchange components (sensor units, primary electronics);

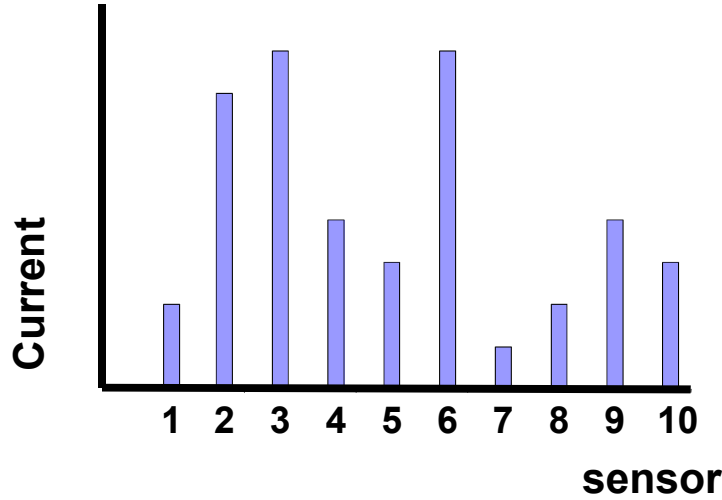
Measurement sequence with e-noses



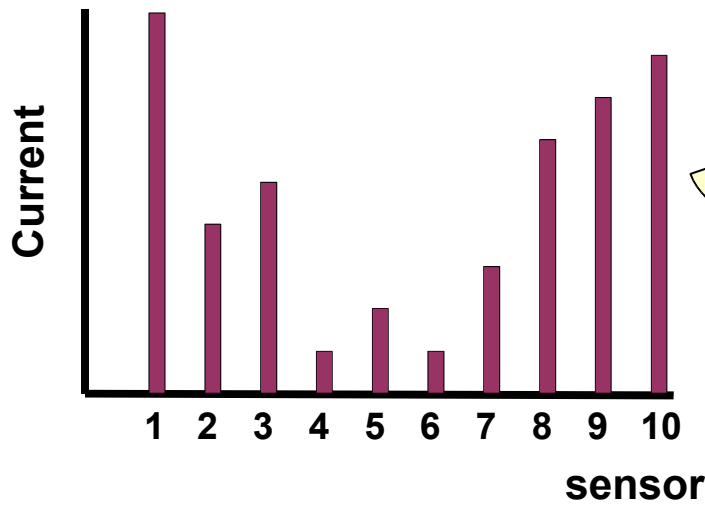
Electronic nose \neq gas monitor

recognition algorithm

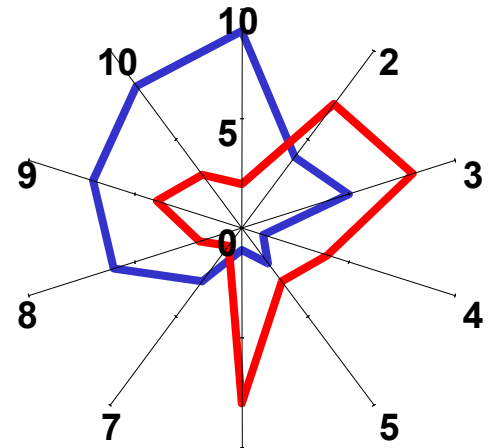
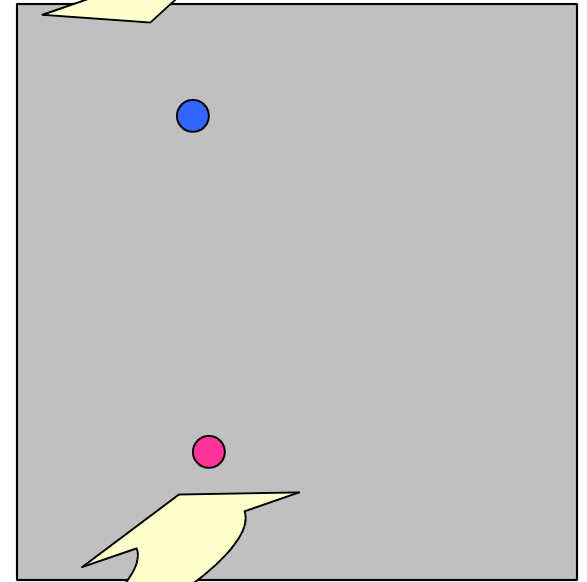
Odour "blue"



Odour "red"



PCA plane

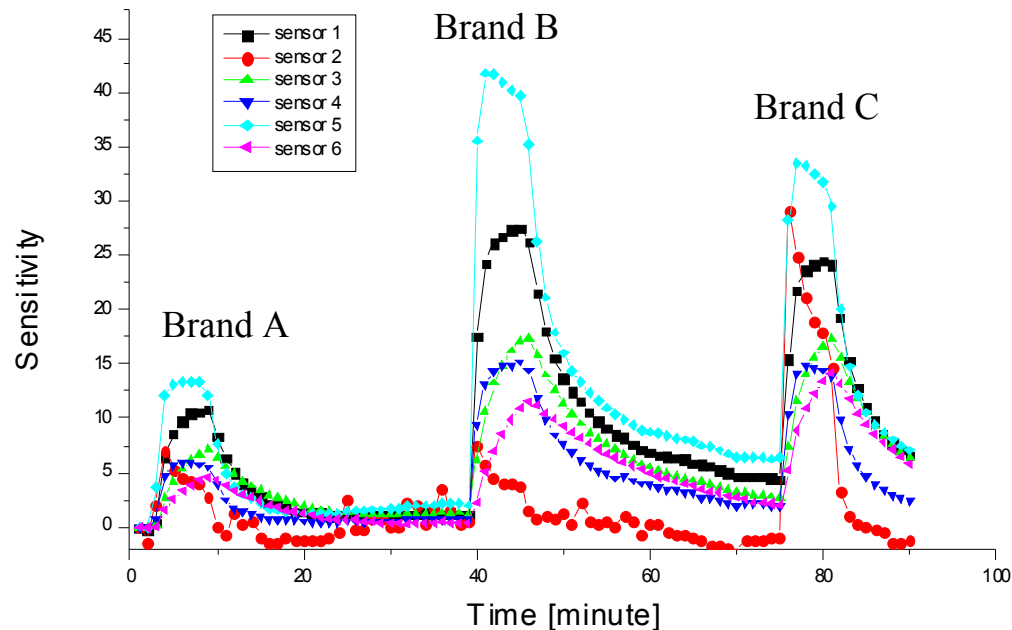


Polar representation = "odour fingerprint"

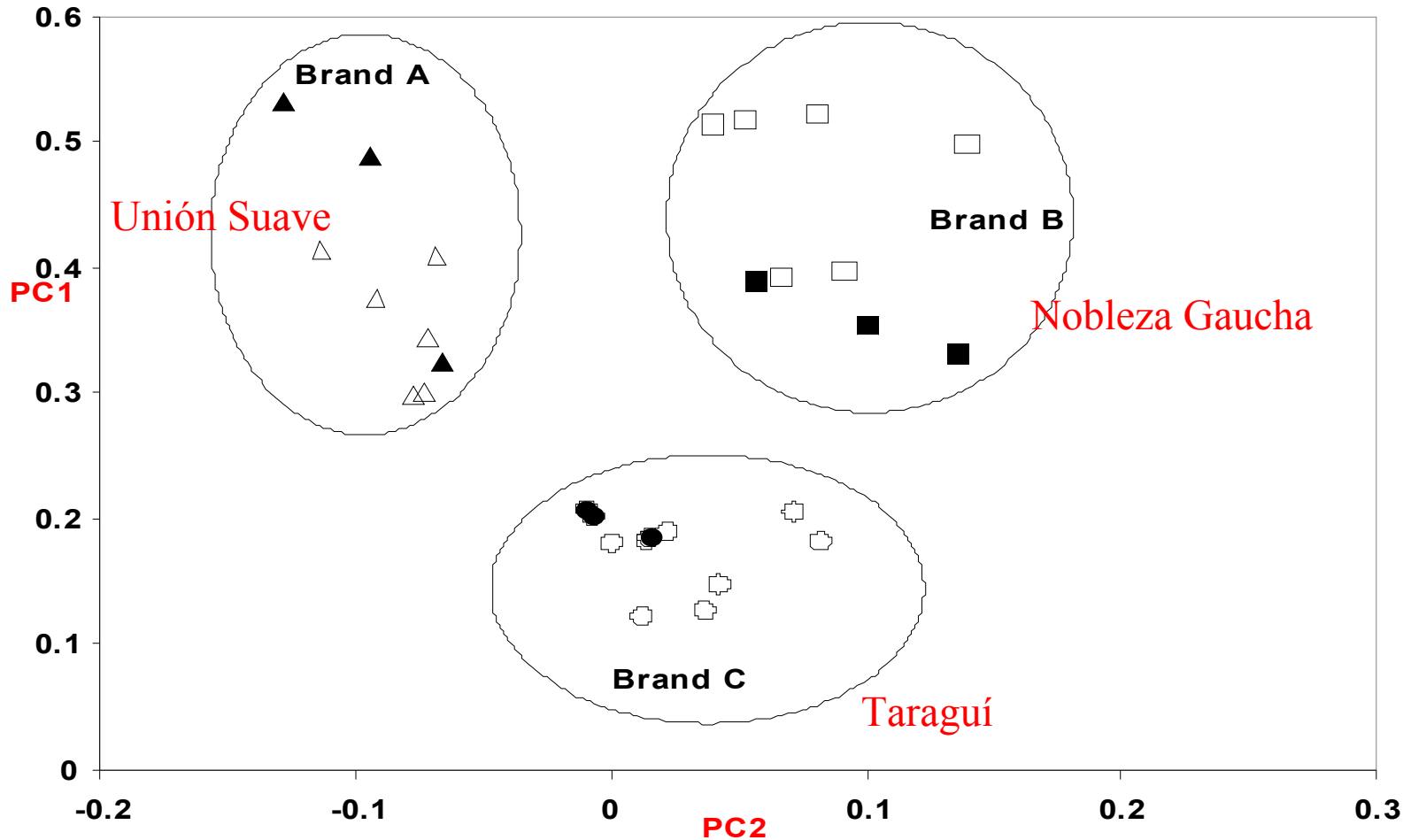
Characterization of different mate brands

- ① For each sample, six cyclic exposures to the odour molecules were repeated.
- ① Each exposure to the mate odour was taken during 3 minutes preceded and followed by an air purge for 20 minutes.
- ① We analyzed the sensitivity defined as: $S = (I - I_{\text{base}}) / I_{\text{base}}$

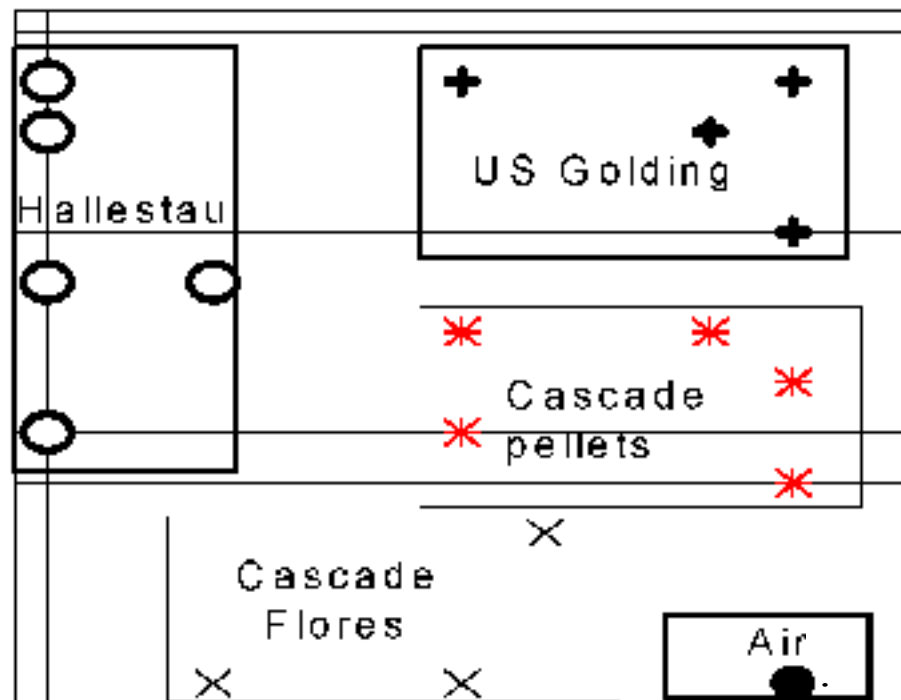
with I (I_{base}) the sensor current in the presence of odour (air).



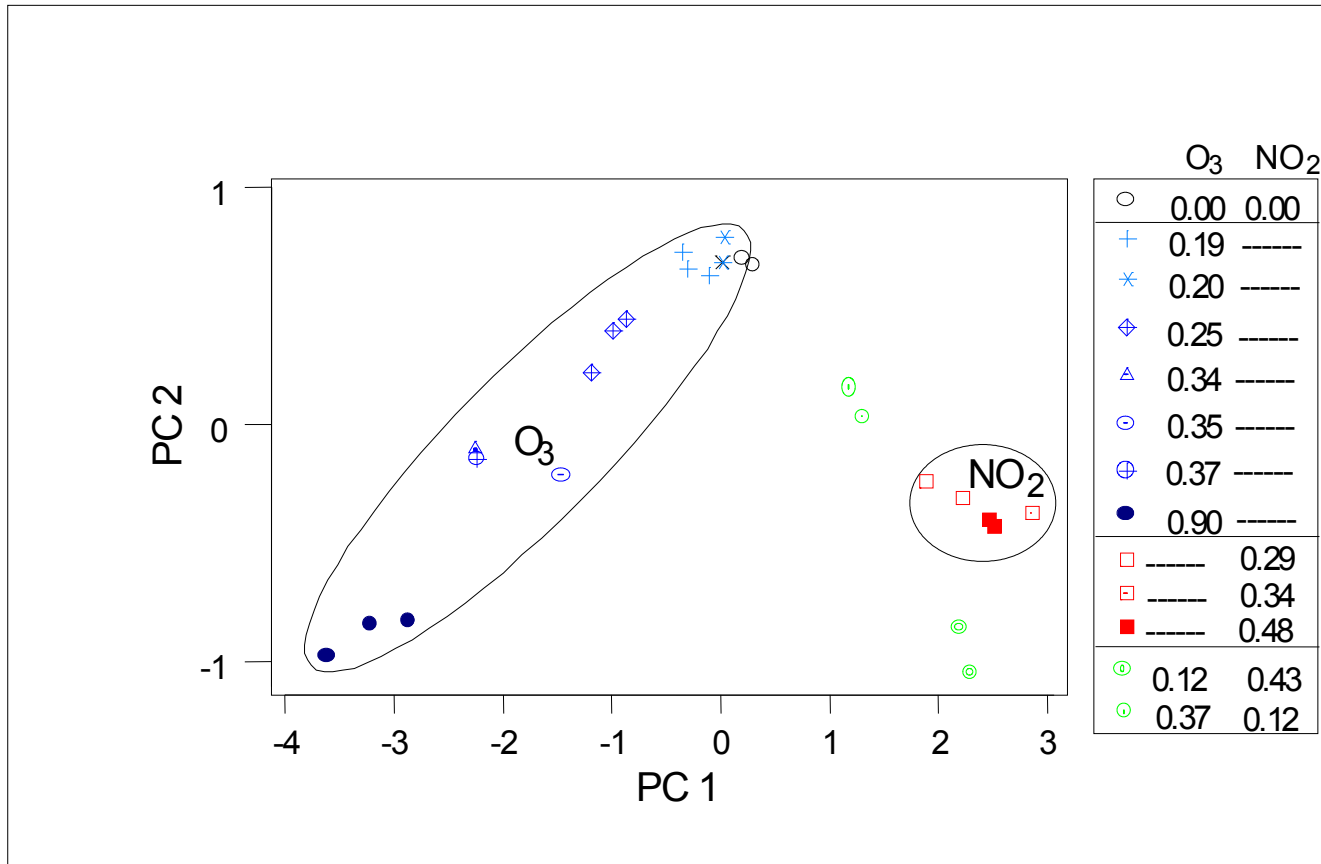
Principal component analysis (PCA) was applied (3). The PCA coefficients were defined using as learning set 18 measurements, 6 for each brand (black points). "Unknown" samples were measured and the corresponding nose response was projected onto the previously defined subspace (red points). The adequacy of the scheme for recognition purposes is evident from the graph.



Self Organizing Maps (SOM) of different samples of hops



Both ozone (O₃) and nitrogen dioxide (NO₂) simultaneously present



Principal components analysis of ozone and NO₂

- ① The presence of either O₃ or NO₂ are clearly discriminated
- ① The nose response is ambiguous if only a PCA analysis is done, when both oxidants are present with similar concentrations.
- ① This result is shared by most types of sensor including electrochemical cells.

Commercial Desktop E- NOSE

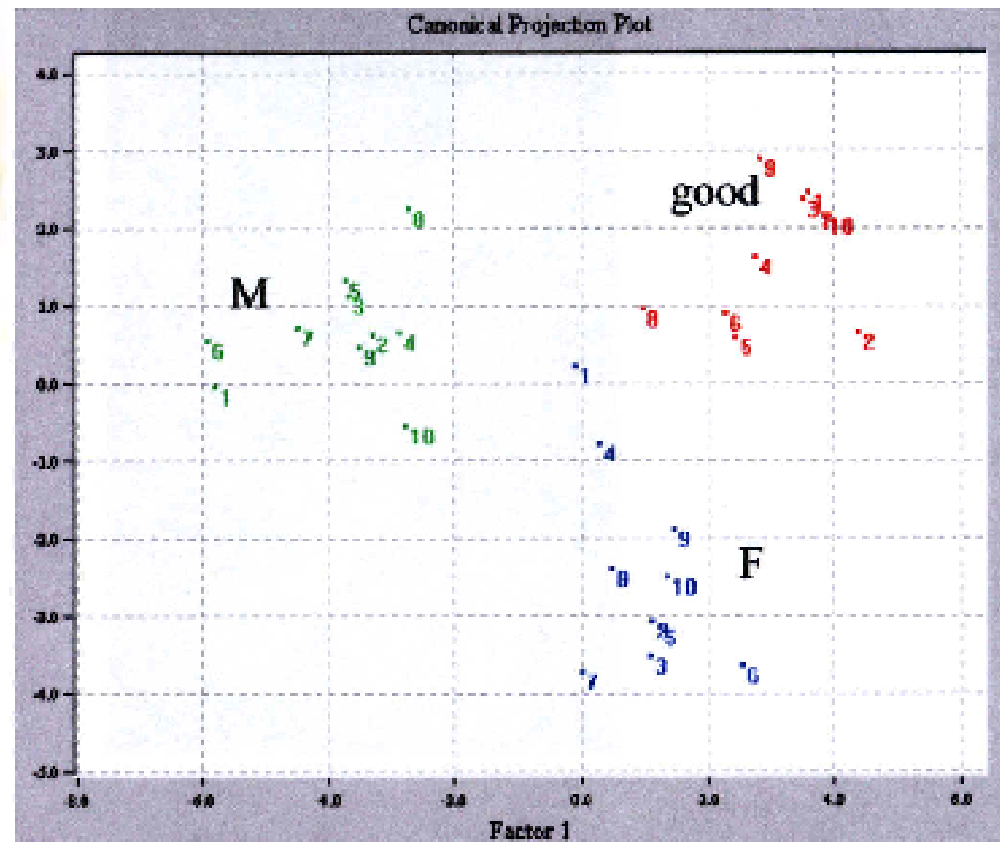




CYRANO
sciences



Commercial Handheld e-nose

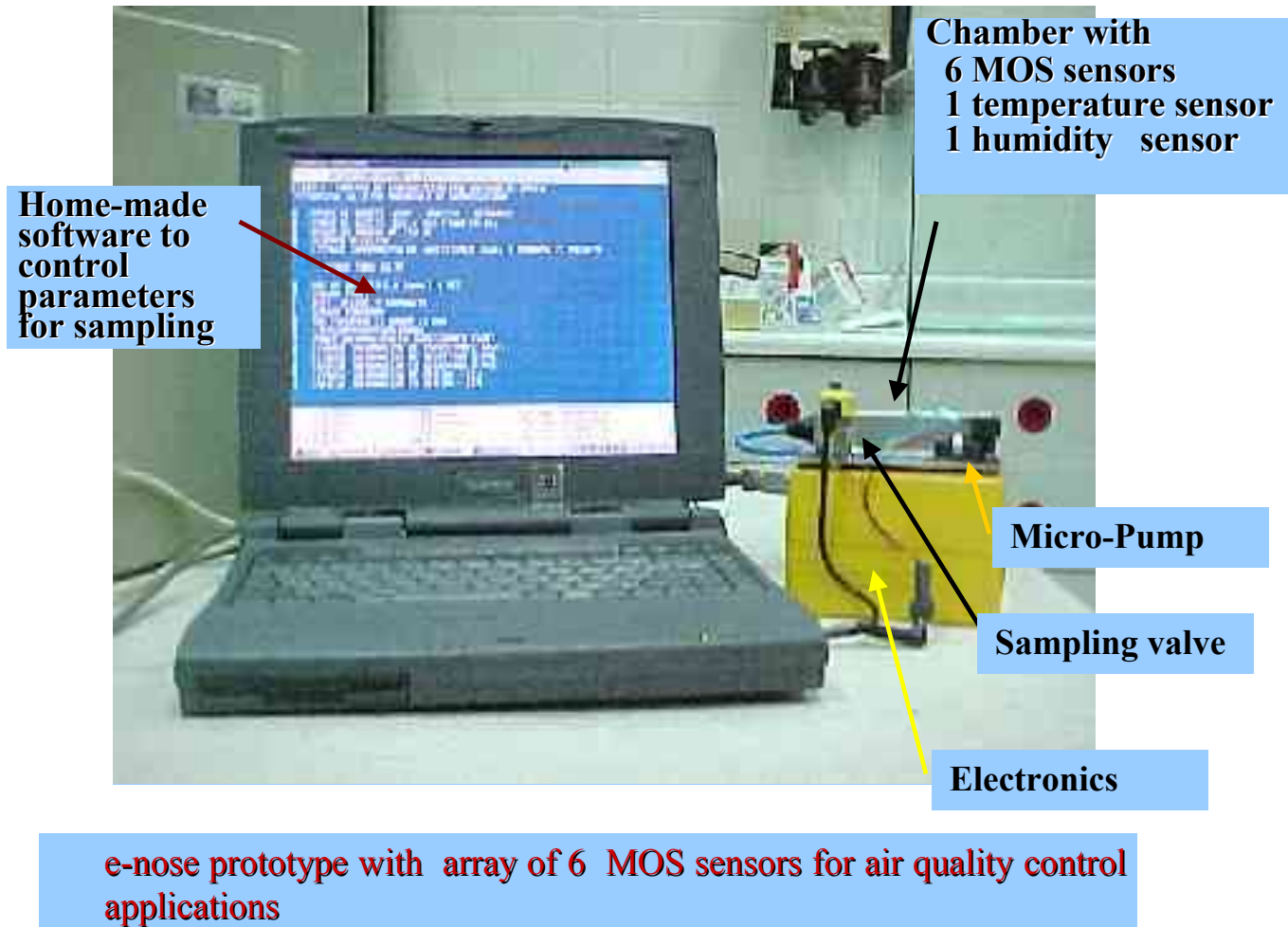


Commercial e-noses

AppliedSensor 



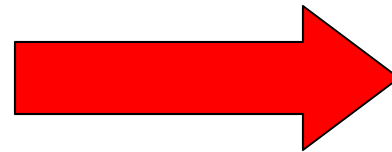
Our First “Lab Prototype” of Electronic Nose



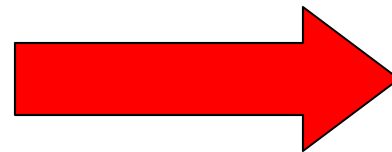
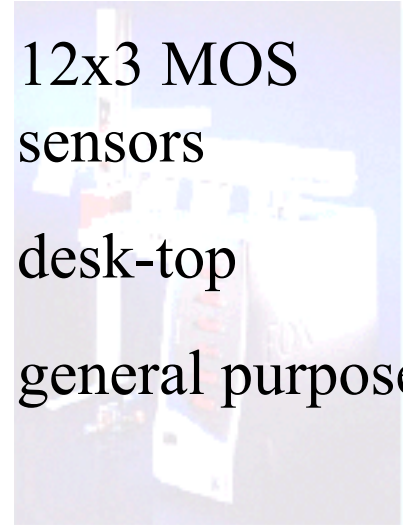
Our development of e-noses under the PID131-2001 contract SECYT-JPA SA



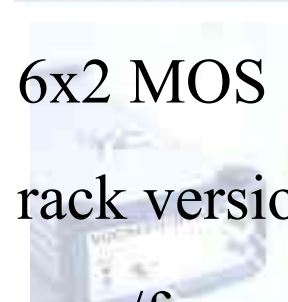
Second “Lab Prototype”
of Electronic Nose (dic-2003)



12x3 MOS
sensors
desk-top
general purpose



6x2 MOS
rack version
one /few product



2 versions of “Pre-competitive Prototype”
to be transferred to the industry JPA SA

Nov. 2004

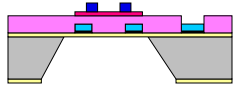
Market prospective

Product Types	1996 Units (millions)	\$ (millions)	2002 Units (millions)	\$ (millions)
Drug delivery systems	1	10	100	1000
Optical switches	1	50	40	1000
Lab on ship	0	0	100	1000
Magneto optical heads	0.01	1	100	500
Projection valves	0.1	10	1	300
Coil on chip	20	10	600	100
Micro relays	0.1	50	100	
Micro motors	0.1	5	2	80
Inclinometers	1	10	20	70
Injection nozzles	10	10	30	30
Anti-collision sensors	0.01	0.5	2	20
Electronic noses	0.001	0.1	0.05	5
TOTAL	33	\$107	1045	\$4,205

Table 4. Worldwide M 3 market size in 1996 and 2002 for **emerging MEMS** product types in \$US millions (Prime Faraday Partnership ,Wolfson School of Mech. And Manuf. Eng., Loughborough University, UK, 2002)

E-nose group challenges and future trends

Single MOS sensor per membrane



CNEA-Citefa



Multiple MOS sensor per membrane (single nose chip sensor). Two concepts in “draft” to be implemented (2005-2006)

Conventional Electronics

CeTAD -Citefa



IC custom made (2006)

PME 109



¿ JPA SA? +...R&D groups



Hand held e-nose
few odours but reliable
(2007)

Summary

- * Develop a MEMS gas sensor and an application , e-nose, with the National Agency of Science and Tech funds and an industrial partner (www.e-nose.com.ar)

- * Start-up with a MEMS activity at CNEA and a cooperation with other R&D Institutes that allows us to develop more interesting devices

- * We received funds for modernizing the lab equipment that will allow us test new gas sensors devices for a new generation of handheld e-noses